Fluorolytic Sol-Gel synthesis: A new approach towards nanoscopic metal fluorides

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The *fluorolytic* sol gel synthesis of nanoscopic metal fluorides has been developed by our group over the past 8 years. The principle consists in the reaction of a metal alkoxide with anhydrous HF in alcoholic solution resulting in an OR against F exchange (*fluorolysis*)

 $M(OR)_n + n HF \longrightarrow MF_n + n ROH$ (1)

resulting in metal fluoride sols which – under appropriate conditions – can be converted into nanoscopic pure Lewis acidic metal fluorides with a very high surface area [1-3].

By combining the *fluorolytic* with the *hydrolytic* sol-gel synthesis this approach can be further extended, thus giving access to metal hydroxide fluorides (eq. 2) or oxide fluorides (eq. 3).

 $M(OR)_n + n - x HF + xH_2O \longrightarrow M(OH)_xF_{n-x} + n ROH$ (2)

 $M(OH)_x F_{n-x}$ $T \longrightarrow MO_{x/2} F_{n-x} + H_2O$

Due to their unique properties, these new materials are of interest for applications in catalysis, optical materials, ceramics, and inorganic-organic hybrid materials. Thus, along with physico-chemical data, examples for catalytic and optical applications will be presented.

reflection [%]

8

9

10

400

500



Image to illustrate the antireflective effect of an MgF_2 coating on glass

Optical data for MgF₂ layers of different thickness.

700

wavelength [nm]

800

600

MgF, on Float 3mm

(3)

4 mm/sec. 5 mm/sec.

6 mm/sec.

7 mm/sec.

8 mm/sec.

9 mm/sec.

Based on nanoscopic metal fluorides even new composite materials with exciting optical, thermal and mechanical properties are accessible.



[1] E. Kemnitz, U. Groß, St. Rüdiger, S.C. Shekar, Angew. Chem. Int. Ed. 42 (2003) 4251;

[2] S. Rüdiger, U. Groß, E. Kemnitz, J. Fluorine Chem. 128 (2007) 353.

[3] S. Rüdiger, E. Kemnitz, Dalton Trans. (2008) 1117.